

- NOTE TAKING:
1. Put the date on the paper.
  2. Fill in all the notes
  3. Add notes from the board, etc.
  4. Add information from the text.



## I EXPLORING PHYSICAL SCIENCE:

### A. That's Science !

1. What is science? \_\_\_\_\_  
\_\_\_\_\_

### B. What is Physical Science?

1. Physical science is the study of \_\_\_\_\_ & \_\_\_\_\_
2. Matter is the " \_\_\_\_\_ " that everything is made of.
3. Energy is the \_\_\_\_\_ to do \_\_\_\_\_.

### C. Branches of Physical Science: (3 main ones)

1. \_\_\_\_\_ - \_\_\_\_\_
2. \_\_\_\_\_ - \_\_\_\_\_
3. \_\_\_\_\_ - \_\_\_\_\_

### D. Physical Science: All Around You:

1. 4 more sciences:

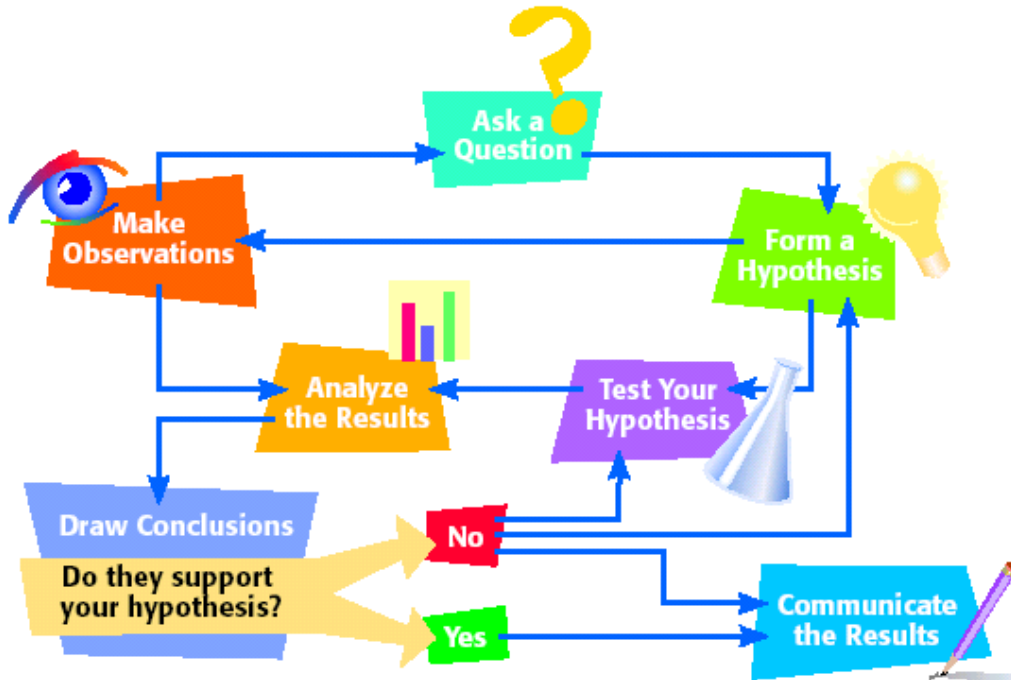
- a) \_\_\_\_\_ - \_\_\_\_\_
- b) \_\_\_\_\_ - \_\_\_\_\_
- c) \_\_\_\_\_ - \_\_\_\_\_
- d) \_\_\_\_\_ - \_\_\_\_\_



## II SCIENTIFIC METHODS:

### A. What are Scientific Methods?

1. They are the \_\_\_\_\_ that scientists answer \_\_\_\_\_ and solve \_\_\_\_\_.



### B. Asking a Question

1. Helps \_\_\_\_\_ the purpose of an \_\_\_\_\_.
2. Usually asked after many \_\_\_\_\_.
3. An observation is any use of the \_\_\_\_\_ to gather \_\_\_\_\_.

### C. Forming a Hypothesis

1. It is a possible \_\_\_\_\_ or \_\_\_\_\_ to a question.
2. A good hypothesis is \_\_\_\_\_.
3. Scientists often make a \_\_\_\_\_ before they test the hypothesis.

**D. Testing the Hypothesis**

1. A controlled experiment \_\_\_\_\_ the results from a control group with an experimental group.
2. Independent variable - the one thing that \_\_\_\_\_ change.
3. Dependent variable- the \_\_\_\_\_ to the independent one.

**D. Analyzing the Results**

1. After you \_\_\_\_\_ & \_\_\_\_\_ data, you must analyze them.
2. You must find out if the results \_\_\_\_\_ the hypothesis.
3. \_\_\_\_\_ & \_\_\_\_\_ are useful.

**E. Drawing Conclusions**

1. \_\_\_\_\_
2. \_\_\_\_\_
3. \_\_\_\_\_



**III SCIENTIFIC MODELS:**

**A. Models in Science**

1. A model is a \_\_\_\_\_ if an object or a system.
2. Three common kinds of scientific models:
  - a. \_\_\_\_\_ - \_\_\_\_\_
  - b. \_\_\_\_\_ - \_\_\_\_\_
  - c. \_\_\_\_\_ - \_\_\_\_\_

**B. Using Models**

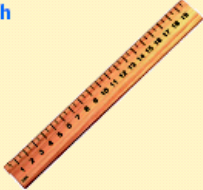



1. They can also be \_\_\_\_\_ to help learn new information.
2. Theory- \_\_\_\_\_
3. Scientific law- \_\_\_\_\_

**A. Tools in Science**

1. One way to collect \_\_\_\_\_ is to take \_\_\_\_\_.
2. Some tools are: \_\_\_\_\_

**B. Making Measurements**

1. SI is the \_\_\_\_\_
2. Two reasons why it is a good system:
  - a. \_\_\_\_\_
  - b. \_\_\_\_\_

Common SI Units		
<b>Length</b> 	<b>meter (m)</b> kilometer (km) decimeter (dm) centimeter (cm) millimeter (mm) micrometer ( $\mu\text{m}$ ) nanometer (nm)	$1 \text{ km} = 1,000 \text{ m}$ $1 \text{ dm} = 0.1 \text{ m}$ $1 \text{ cm} = 0.01 \text{ m}$ $1 \text{ mm} = 0.001 \text{ m}$ $1 \mu\text{m} = 0.000001 \text{ m}$ $1 \text{ nm} = 0.000000001 \text{ m}$
<b>Volume</b> 	<b>cubic meter (<math>\text{m}^3</math>)</b> cubic centimeter ( $\text{cm}^3$ ) liter (L) milliliter (mL)	$1 \text{ cm}^3 = 0.000001 \text{ m}^3$ $1 \text{ L} = 1 \text{ dm}^3 = 0.001 \text{ m}^3$ $1 \text{ mL} = 0.001 \text{ L} = 1 \text{ cm}^3$
<b>Mass</b> 	<b>kilogram (kg)</b> gram (g) milligram (mg)	$1 \text{ g} = 0.001 \text{ kg}$ $1 \text{ mg} = 0.000001 \text{ kg}$
<b>Temperature</b> 	<b>Kelvin (K)</b> Celsius ( $^{\circ}\text{C}$ )	$0^{\circ}\text{C} = 273 \text{ K}$ $100^{\circ}\text{C} = 373 \text{ K}$

3. Length: \_\_\_\_\_  
 Tool: \_\_\_\_\_ Units: \_\_\_\_\_
4. Mass: \_\_\_\_\_  
 Tool: \_\_\_\_\_ Units: \_\_\_\_\_
5. Volume: \_\_\_\_\_  
 Tool: \_\_\_\_\_ Units: \_\_\_\_\_
6. Temperature: \_\_\_\_\_  
 Tool: \_\_\_\_\_ Units: \_\_\_\_\_

### C. Metric Ruler:

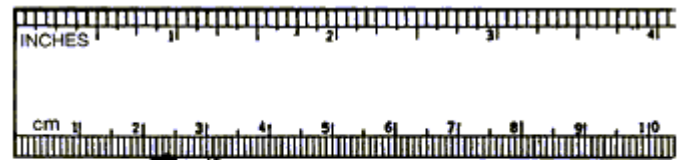
1. When reading the metric ruler, remember:

a. There are \_\_\_\_ mm in 1 cm.

b. The leaf at right is :

\_\_\_\_\_ mm or \_\_\_\_\_ cm

c. Be careful, the leaf starts at the 1 cm mark.



Leaf A

### D. Triple Beam Balance:

You must move the sliders until they are locked into one of the notches in the bar

Each of the small lines on the front bar show 0.1 grams of mass

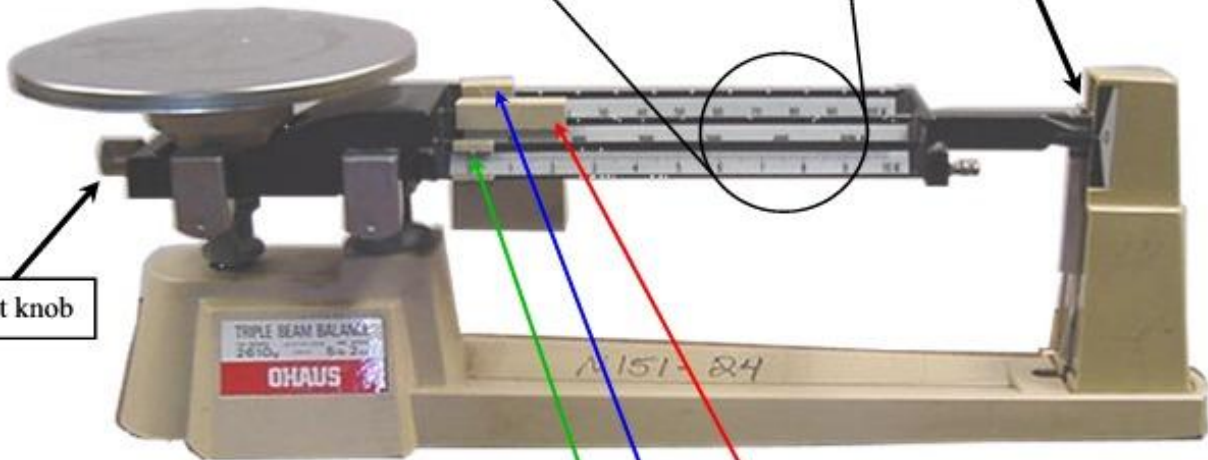
When the balance is empty the pointer should line up with the zero mark. If it does not, turn the adjustment knob at the other end of the balance.

adjustment knob

first - move the middle slider

second - move the back slider

third - move the front slider



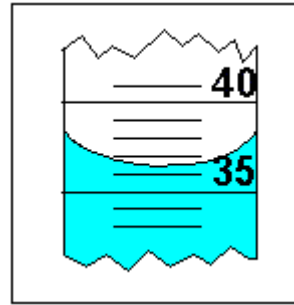
## E. Graduated Cylinder:

1. When measuring liquid, hold the graduated cylinder so you are at \_\_\_\_\_ level with the level of the liquid.

2. Notice the liquid is curved downward:

This is called the \_\_\_\_\_

The reading on the picture at right is \_\_\_\_\_ mL.

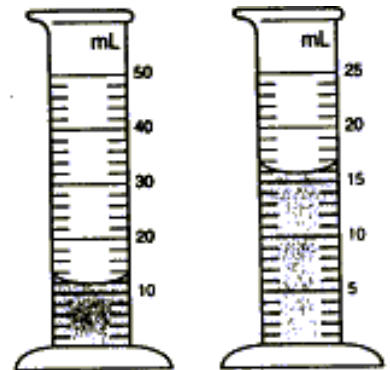


3. Sometimes graduated cylinders have lines for every mL and sometimes for every two mL. Look at the cylinders below:

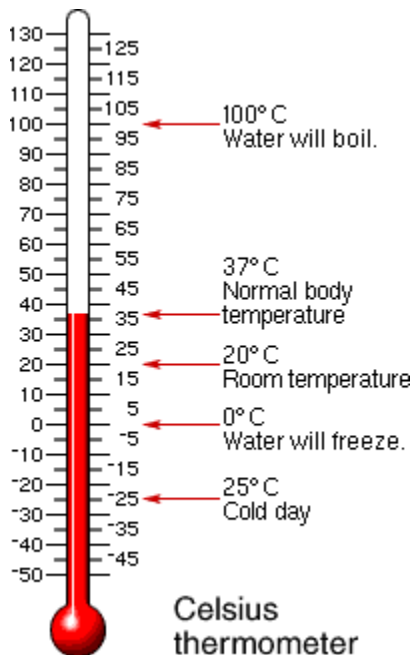
a. What is the volume of liquid for the graduated cylinder on the...

LEFT: \_\_\_\_\_

RIGHT: \_\_\_\_\_



## F. Thermometer:



On the Celsius Scale:

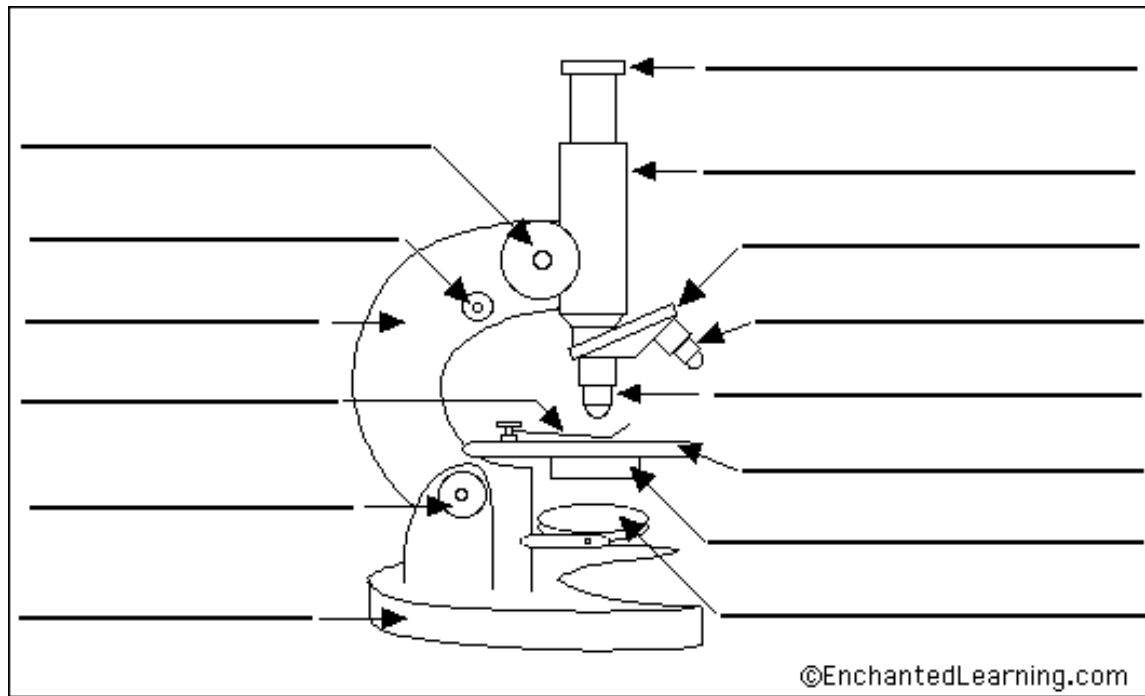
1. Water boils at \_\_\_\_\_

2. Water freezes at \_\_\_\_\_

3. Normal body temperature is \_\_\_\_\_



**1. Diagram:**



**2. Parts:**

a. Eyepiece: \_\_\_\_\_  
 \_\_\_\_\_

b. revolving nosepiece: \_\_\_\_\_  
 \_\_\_\_\_

c. objectives: \_\_\_\_\_  
 \_\_\_\_\_

d. stage: \_\_\_\_\_

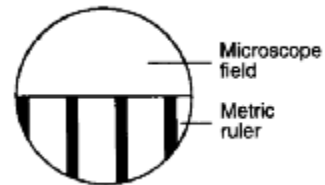
e. diaphragm: \_\_\_\_\_  
 \_\_\_\_\_

f. coarse focus knob: \_\_\_\_\_  
 \_\_\_\_\_

g. fine focus knob: \_\_\_\_\_  
 \_\_\_\_\_

3. Image:

- a. Due to the way the lenses bend the light waves, the image you see is \_\_\_\_\_ and \_\_\_\_\_
- b. When you look into the microscope, you see a \_\_\_\_\_ area of light. This is called the \_\_\_\_\_ of \_\_\_\_\_
- c. On low power, the field of view is approximately \_\_\_\_\_ mm in diameter.



4. Rules for using the microscope:

- b. \_\_\_\_\_  
\_\_\_\_\_
- c. \_\_\_\_\_  
\_\_\_\_\_
- d. \_\_\_\_\_  
\_\_\_\_\_
- e. \_\_\_\_\_  
\_\_\_\_\_
- f. \_\_\_\_\_  
\_\_\_\_\_
- g. \_\_\_\_\_  
\_\_\_\_\_

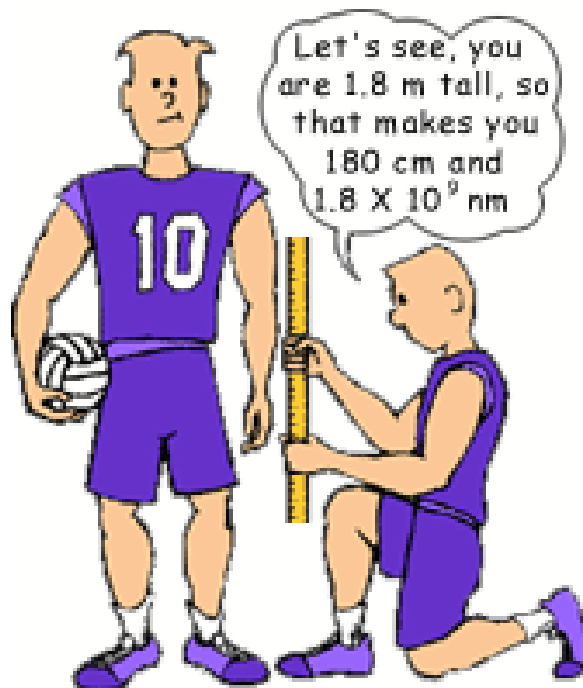
4. When you are finished with the microscope, do the following:

- a. \_\_\_\_\_
- b. \_\_\_\_\_
- c. \_\_\_\_\_



1. The metric system is easier to use because:

- a. It is based on \_\_\_\_\_
- b. You only have to \_\_\_\_\_  
the decimal point \_\_\_\_\_ or  
\_\_\_\_\_.



2. The prefixes used the most are:

Kilo: \_\_\_\_\_

Deka: \_\_\_\_\_

Deci: \_\_\_\_\_

Centi: \_\_\_\_\_

Milli: \_\_\_\_\_

3. Look at the PREFIX table below:

KILO	HECTO	DEKA	Gram Liter Meter	DECI	CENTI	MILLI
1000	100	10		.1	.01	.001

- a. When you convert, use this table to know which way to move the decimal point and how many places.

4. Example:

- a. If you have 23 centimeters, how many millimeters is that?

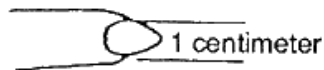
Start at the Centi box in the table, move to the Milli box.

You went one place to the right, so the decimal point goes

one place to the right. The answer is: \_\_\_\_\_

- a. 1200 grams = \_\_\_\_\_ kilograms
- b. 35 kilograms = \_\_\_\_\_ hectograms
- c. 4.5 kilograms = \_\_\_\_\_ grams
- d. 13 millimeters = \_\_\_\_\_ centimeters
- e. 13 millimeters = \_\_\_\_\_ meters
- f. 54 centimeters = \_\_\_\_\_ meters
- g. 76 decimeters = \_\_\_\_\_ meters
- h. 85 centimeters = \_\_\_\_\_ decimeters
- i. 4800 meters = \_\_\_\_\_ kilometers
- j. 5734 centimeters = \_\_\_\_\_ hectometers

**1 centimeter (or 1 cm) = the width of  
some part of your smallest finger or  
fingernail**



**1 kilometer (or 1 km) = a little more than  
half a mile (pronounced KILL-oh-meet-ur not kill-AHM-  
it-ur)**

